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Section 3. Terms in this set (7) mole. The SI base unit used to measure the amount of a substance whose number of particles is the same as the number of atoms of carbon in exactly 12 grams of carbon ...

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measure the amount of....  $6.022 \times 10^{23}$  of anything. the mass in grams of one mole of a subs.... the periodic table - an elements mass l.... mole.

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formula mass of one molecule is  $x$

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the word mole represents a number of things  $6.022 \times 10^{23}$  of them but does not by itself specify what they are the chemical entities can be atoms

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