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Experiment A5- Reactions - 1 EXPERIMENT A5 TYPES OF ...

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Question: Chemical Kinetics Is The Study Of The Rate Of Chemical Reactions And The Factors That Influence That Rate. In This Experiment, You Will Be Studying The Kinetics Of The Reaction Of Iodide Ion With Bromate Ion In Acidic Conditions: 6 I-(aq) + BrO3-(aq) + 6 H+(aq) ----> 3 I2(aq) + Br-(aq) + 3 H2O(l) The Reaction Rate Depends On The Concentration Of Reactants ...

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EXPERIMENT A7 continued 5. Organizing Ideas Write the equation for the reaction that occurred when you heated hydrated MgSO4 in this experiment. Use the letter n to represent the number of moles of water driven off per mole of anhydrous magnesium sulfate. 6. Organizing Conclusions Using your answers to Calculations items 2,

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Can somebody kindly explain what this experiment is about? A5 is used for this experiment. And how to derive the moment of inertia of the top plank with tee attached to its free-end, as well as determining the center of gravity distance from the hinge?

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